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1 Solid lipid nanoparticles loaded with lipoyl-
2 memantine codrug: preparation and characterization

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1 **Abstract**

2 Solid lipid nanoparticles (SLNs) are considered very attractive drug-delivery systems (DDS) able to
3 enhance the efficacy of some therapeutic agents in several pathologies difficult to treat in a conventional
4 way. Starting from these evidences, this study describes the preparation, physicochemical characterization,
5 release, and *in vitro* cytotoxicity of stealth SLNs as innovative approach to improve solubility and
6 absorption through the gastrointestinal tract of Lipoyl-Memantine (LA-MEM), a potential anti-Alzheimer
7 codrug.

8 Physico-chemical properties of LA-MEM loaded SLNs have been intensively investigated. Differential
9 scanning calorimetry (DSC) was used to clarify the state and crystalline structure of the formulation. The
10 results obtained from particles size analysis, polydispersity (PDI), and zeta potential measurements allowed
11 the identification of the optimized formulation, which was characterized by a drug-lipid ratio 1:5, an
12 average intensity diameter of 170 nm, a PDI of 0.072, a zeta potential of -33.8 mV, and an entrapment
13 efficiency of 88%. Moreover, *in vitro* stability and release studies in both simulated gastric fluid (SGF) and
14 simulated intestinal fluid (SIF), and preliminary *in vitro* cytotoxicity studies revealed that LA-MEM loaded
15 SLNs could represent potential candidate for an *in vivo* investigation as DDS for the brain since it resulted
16 devoid of citotoxicity and able to release the free codrug.

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18 **Keywords:** Solid lipid nanoparticles, Alzheimer's disease, Memantine, Lipoic acid, Codrug.

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1 **1. Introduction**

2 Nanotechnological approaches are often used to improve the pharmacokinetic profile of drugs such as poor
3 gastro-intestinal absorption, low drug solubility, and rapid metabolism but also to ensure an efficient CNS
4 delivery of many compounds. Among various colloidal systems, SLNs have roused special interest due to
5 many features that make them particularly intriguing as alternative carriers in the field of drug delivery
6 (Kanwar et al., 2009). In combination with the narrow sizes, sterical properties obtained in stealth SLNs
7 result particularly advantageous to increase the blood circulation time of the particulate, thus reducing RES
8 uptake and extending the contact time between the BBB and the drug, which can therefore be caught in the
9 brain when the target is represented by CNS (Uner and Yener, 2007). Such properties can improve
10 bioavailability of drugs through several mechanisms: 1) augmenting the drug solubility and permeability,
11 2) overcoming the first-pass effect and the P-gp efflux, and 3) enhancing the stability in the GI tract
12 (Esposito et al., 2008). Several drug-loaded SLN formulations in brain targeting and delivery showed a
13 very low cytotoxicity or resulted to be not-cytotoxic (Patel et al., 2013). However, the biocompatibility
14 assessment seemed to be tough due to the variable composition of SLN formulations that differs in nature,
15 percentage of lipids and loaded molecules (Silva et al., 2012).

16 Recently, our research has been focused on the development of innovative therapeutic strategies to reduce
17 the disease progression and improve the quality of life of patients suffering from Parkinson's (PD) (Di
18 Stefano et al., 2009) and Alzheimer's (AD) Diseases (Cacciatore et al., 2012, 2013; Sozio et al., 2009,
19 2013a). To date, AD treatments mainly exploit a symptomatic approach based on the use of cholinesterase
20 inhibitors for patients with mild to moderate AD, and N-methyl-D-aspartate (NMDA) receptor antagonists,
21 such as memantine (3,5-dimethyl-1-adamantanamine, MEM), for patients with moderate to severe AD
22 (Anand et al., 2014; Parihar et al., 2004). According to the multifactorial hypothesis of AD, glutamate
23 excitotoxicity, oxidative stress (OS), depletion of antioxidant defence systems, metal dyshomeostasis,
24 and protein misfolding and aggregation are considered the main causes of neuronal loss in selected brain
25 regions (Di Stefano et al., 2011). Taking into account the role played by OS and over-stimulation of
26 glutamate NMDA receptors in causing neuronal damage, we previously reported the synthesis of Lipoyl-
27 Memantine (LA-MEM) (Figure 1), a new potential anti-Alzheimer codrug obtained by joining MEM with a
28 natural neuroprotective agent, (R)- α -lipoic acid (LA) (Sozio et al., 2013b). Despite of its good capabilities

1 to inhibit A β (1–42) aggregation and scavenge free radicals, LA-MEM showed poor water solubility (0.010
2 mg/mL) to permit an oral administration.

3 This paper is focused on the preparation and characterization of a new SLN formulation containing LA-
4 MEM codrug to increase its solubility in g.i. fluids and favor its intestinal absorption. High lipophilic
5 compounds, such as LA-MEM (LogP = 4.20), have often been selected to be incorporated into SLNs due to
6 their high drug loading and entrapment efficiency (Al Haj et al., 2008). The new formulation was
7 characterized evaluating particle size, zeta potential, surface morphology, drug loading capacity, drug
8 encapsulation efficiency, and stability in simulated gastric (SGF) and intestinal (SIF) fluids. Few studies
9 are reported in literature about the potential toxicity of SLN formulations (Nassimi et al., 2010), thus *in*
10 *vitro* cytotoxicity of LA-MEM loaded SLNs was assessed against mouse N2a neuroblastoma (NB) and
11 primary human whole blood (PHWB) cells using MTT and lactate dehydrogenase (LDH) assays,
12 respectively. Total antioxidant capacity (TAC) and total oxidative stress (TOS) levels were also determined
13 to valuate oxidative alterations.

14 **2. Material and methods**

15 *2.1. Chemicals*

16 LA-MEM was synthesized as we previously reported (Sozio et al., 2013b). Stearic acid, Brij 78 and
17 Acetone were purchased by Sigma Aldrich (St. Louis, MO, USA). Sucrose was obtained from Alfa Aesar
18 (MA, USA), Cremophor[®] ELP was obtained from BASF (Ludwigshafen, Germany). NaCl, NaOH,
19 KH₂PO₄ and all other solvents of chromatographic grade were obtained from Fisher (Loughborough, UK).
20 MilliQ water was produced by a MilliQ Direct-Q UV3 Millipore system (Merck, Darmstadt, Germany).
21 Chromafil[®] PET-120/25 filters (Duren, Germany) were used for all the formulations.

22 *2.2. Preparation of empty and LA-MEM loaded stealth Solid Lipid Nanoparticles*

23 SLNs were prepared according to a slight modification of the emulsification-evaporation-solidifying
24 method previously reported (Sjostrom et al., 1995; Shahgaldian et al., 2003). For the blank SLN, stearic
25 acid (SA) (33.36 mg) was completely dissolved in 8 mL of acetone under mild magnetic stirring at 40 \pm 2
26 $^{\circ}$ C. Concurrently, a solution of surfactant (Brij 78) in deionized water, with different lipid:surfactant ratio
27 $^{\circ}$ C.

1 (1:0.4 and 1:1), was prepared and heated at 75 ± 2 °C. The organic phase was added dropwise to the
2 aqueous phase under mechanical stirring (1000 rpm) to obtain the lipid emulsion which was subsequently
3 concentrated to 11.2 mL evaporating the solvents mixture at 75 ± 2 °C. The translucent emulsion system
4 was quickly added to 11.2 mL of cold deionized water on ice bath under mechanical stirring (1000 rpm).
5 The obtained solidified nanoparticles suspension was adjusted to total 25 mL with cold distilled water. The
6 drug-loaded nanoparticles were prepared with the same procedure adding LA-MEM to the organic phase
7 with different drug:lipid ratio (1:2.5, 1:5, 1:10, 1:20). The final suspension was filtered with 1.20 μ m filters
8 to remove the untrapped drug.

9

10 *2.3. Dynamic Light Scattering and Zeta Potential measurement*

11 Intensity mean hydrodynamic size, polydispersity index (PDI), and Zeta potential of the particles were
12 measured on a Malvern Zetasizer-NanoZS with a He-Ne laser with a wavelength of 632.8 nm. The
13 measurements were carried out at a scattering angle of 173° using disposable sizing cuvette and keeping the
14 temperature at 25 °C throughout the experiments. All measurements were repeated three times ($n = 3$) and
15 the results are given as the effective diameter, also called Z-average diameter, and the PDI as a parameter
16 of the particle size distribution. The measurements were performed at 25 °C, in triplicate ($n = 3$) and the
17 average values were calculated. The electrophoretic mobility was determined in an aqueous medium with a
18 Smoluchowski approximation.

19

20 *2.4. Lyophilization of SLNs*

21 SLNs dispersed in water were subjected to lyophilization to obtain dried formulations. Two different
22 cryoprotectants were tested to avoid coalescence of SLNs (Shwarz et al., 1997; Cavalli et al., 1997), 0.8 mL
23 of sucrose (5% w/v, 10% w/v) and 0.8 mL of trehalose (5% w/v, 10% w/v) aqueous solutions were mixed
24 with 2 mL of SLNs water suspension; the mixture was then frozen at -20 °C overnight. The lyophilization
25 process was carried out using a VirTis Advantage Freeze Dryer. The freeze-drying protocol employed was
26 the following:

27 Freezing: Shelf temperature -38 °C for 120 minutes.

1 Primary drying (sublimation): Step 1: Shelf temperature -30 °C, vacuum 200 mTorr for 120 minutes; Step
2 2: Shelf temperature -10 °C, vacuum 200 mTorr, for 120 minutes. Secondary Drying (desorption): Shelf
3 temperature: 0 °C, vacuum 200 mTorr, duration 20 hours. Post heat: Shelf temperature: 10 °C, vacuum:
4 200 mTorr, duration 240 minutes.

5

6 *2.5. Morphology evaluation by transmission electron microscopy*

7 Morphological examination of the prepared SLNs was performed using a transmission electron microscope
8 (TEM). A drop of SLNs suspension was placed on a copper grid and stained with 1% (w/v) of uranyl
9 acetate. The sample was viewed under TEM (FEI CM120 BioTwin instrument). Images were recorded
10 using an AMT 5 Mp Digital camera.

11

12 *2.6. Differential Scanning Calorimetry*

13 Differential Scanning Calorimetry (DSC) was performed on DSC Q2000 (TA instruments, LLC, USA).
14 The instrument was calibrated with indium ($T_m = 156.6$, $\Delta H_f = 28.71$ J/g) according to the manufacturer
15 instructions. Nitrogen was used as purge gas with a flow rate of 50 mL/min for all the experiments. TA
16 aluminium pans and lids were used. Data were analysed using TA Instruments Universal analysis 2000. A
17 heating rate in the range of 10 °C/min and 100 °C/min was employed. The samples analysed were stearic
18 acid, LA-MEM, Brij 78, a physical mixture of LA-MEM and stearic acid, blank, and drug-loaded SLNs.

19

20 *2.7. Drug loading capacity, encapsulation efficiency, and yield of production*

21 The amount of encapsulated LA-MEM was quantified by dissolving an aliquot of the nanoparticles
22 suspension (100 µL) in 900 µL of methanol. The obtained solution was analyzed by high performance
23 liquid chromatography (HPLC). Chromatographic analyses were performed on an HPLC system equipped
24 with a quaternary pump, an auto sampler, and a diode array detector. Analyses were carried out using a
25 Waters, X-Terra RP18 column (4.5 x 150 mm, 5 µm pore size). The injection volume was 20 µL and the
26 analyte was eluted at a flow rate of 1 mL/min for an isocratic elution period of 13 min. The mobile phase
27 was composed of 79% v/v ACN and 21% v/v Water with 0.1% v/v TFA. The column was thermostated at

1 25 °C and the detection was carried out by monitoring the absorbance signals at 215 nm. HPLC was
2 calibrated with standard solutions of LA-MEM ranging from 7.85 to 250 µg/mL.

3 The drug loading capacity (DLC%), drug entrapment efficiency (EE%), and yield of nanoparticles
4 (Recovery%), were calculated by the following equations, respectively:

5
$$\text{DLC\%} = (\text{Weight of Codrug in SLNs} / \text{weight of SLNs}) \times 100$$

6
$$\text{EE\%} = (\text{Actual Codrug content} / \text{Theoretical Codrug content}) \times 100$$

7
$$\text{Recovery\%} = (\text{Weight of SLNs Recovered} / \text{Weight of Lipid, Surfactant and Codrug Used Initially}) \times 100.$$

8

9 *2.8. In vitro release studies in Simulated Gastric and Intestinal Fluids*

10 *In vitro* release studies were performed using the dialysis bag method (Verger et al., 1998). Blank SGF at
11 pH 1.2 with 0.1% (v/v) Cremophor ELP and blank SIF at pH 6.8 with 0.1% (v/v) Cremophor ELP were
12 used as dissolution media. In Tables A.1-2 (see appendices) are reported the materials used to prepare the
13 fluids according to the USP. The dialysis bag (molecular weight cutoff 12400 Da) could retain
14 nanoparticles and allow the diffusion of free drug into dissolution media (Luo et al., 2006). The bag was
15 soaked in deionized water for 12 h before use. In detail, 10 mL of drug-loaded SLNs were poured into the
16 dialysis bag, then placed in 400 mL flasks where 200 mL of dissolution medium was added. The ratio
17 drug:fluids used was 1:20. The flasks were placed into a thermostatic shaker (Gallenkamp shaking
18 incubator) at 37 °C and 100 rpm. At specific time intervals (0, 15, 30, 45, 60, 75, 90, and 120 min for SGF,
19 and 0, 15, 30, 60, 90, 120, 180, and 240 min for SIF) aliquots (100 µL) of the suspension were collected
20 from the bags and assayed for the codrug content; 100 µL of the suspension containing SLNs were
21 dissolved in 900 µL of MeOH and analyzed by the HPLC method mentioned above. The percentage of
22 codrug released from the SLNs at each time point was calculated by the following equation:

23
$$\% \text{ released codrug} = 100 - (\% \text{ remaining codrug}).$$

24 Sink conditions were maintained throughout the experiment. All the operations were carried out in
25 triplicate.

26

27 *2.9. In vitro stability of SLNs in Simulated Gastric and Intestinal Fluids*

1 Stability studies were carried out in blank SGF pH 1.2, blank SIF pH 6.8, and in both fluids with pepsin
2 and pancreatin, respectively. 0.1 mL of SLNs suspension were incubated with 1.9 mL of fluids in a
3 thermostatic shaker at 37 °C and 100 rpm. Samples were withdrawn at 0, 15, 30, 60, 90, 120 min for SGF
4 and at 0, 15, 30, 60, 90, 120, 180, 240 for SIF. The samples in the presence of enzymes were centrifuged
5 for 5 min at 10000 rpm and 15 °C to remove pancreatin aggregates from the medium. The particles size and
6 PDI after incubation were determined by DLS on a Malvern Zetasizer-NanoZS, as reported above.
7 A one month preliminary stability study was conducted comparing 10% (w/v) Sucrose blank and drug-
8 loaded SLNs with those stored as aqueous suspensions. The vials were stoppered and crimped for storage.
9 The SLNs were stored at 4 ± 2 °C. Lyophilized SLNs were reconstituted with deionized water and all the
10 samples were tested for changes in the particles size, PDI, and entrapment efficiency. The free drug was
11 removed by filtration with 1.20 µm polyester filter.

12

13 *2.10 Cytotoxicity studies*

14 *2.10.1. Blood samples cell lines*

15 Blood samples were obtained by vein puncture from three healthy, man, non-smoking donors. PHWB
16 cultures were set up according to a slight modification of the protocol described by Evans and O’Riordan
17 (1975). The heparinized blood (0.5 mL) was cultured in 6 mL culture medium (Chromosome Medium B,
18 Biochrom®) with 5 µg/L of phytohemagglutinin (Biochrom®). The mouse N2a-neuroblastoma cell line was
19 obtained from Turkey FMD Institute, Ankara, Turkey. Prior to the experiments, cells were thawed and
20 grown in tissue culture flasks as a monolayer in Dulbecco modified eagles medium (DMEM; Sigma-
21 Aldrich®), supplemented with 1% glutamine, 0.5% penicillin/streptomycin, and 10% fetal bovine serum
22 (PAN Biotech®) at 37 °C in a humidified (95%) incubator with CO₂ (5%). The cultured cells were
23 trypsinised with trypsin/EDTA for a maximum of 5 min and seeded with a sub cultivation ratio of 1:3-1:8.
24 Different concentrations (10 and 100 µM) of SLNs formulations (SLNs, SLNs-LA, SLNs-MEM and SLNs-
25 LA-MEM) were added to the cells. After applications of SLNs formulations, the cultures were incubated
26 for 24 h at 37 °C to adjust body conditions. Untreated cells were used along with H₂O₂ (50 µM, Merck®) as
27 a positive cytotoxic control in all assays.

28

1 2.10.2. *MTT assay*

2 Viability of cells was spectrophotometrically assessed by measuring the formation of formazan crystals via
3 MTT commercial kits (Cayman Chemical[®]). At the end of the experiment, the neurons were incubated with
4 0.7 mg/mL MTT for 30 min at 37 °C. After washing, the blue formazan was extracted from cells with
5 isopropanol/formic acid (95:5) and photometrically determined at 560 nm. The density of formazan formed
6 in untreated cells was taken as 100% viability.

7

8 2.10.3. *LDH assay*

9 Lactate dehydrogenase (LDH) released from damaged cells in culture medium was quantified by using
10 LDH assay kit (Cayman Chemical[®]). A total of 100 µL of cell medium was used for LDH analysis.
11 Released LDH catalyzed the oxidation of lactate to pyruvate with simultaneous reduction of NAD⁺ to
12 NADH. The rate of NAD⁺ reduction was measured as an increase in absorbance at 490 nm, and it resulted
13 directly proportional to LDH activity in the cell medium.

14 2.10.4. *TAC and TOS analyses*

15 Following homogenization, intracellular levels of total antioxidant capacity (TAC) and total oxidant status
16 (TOS) were determined by commercially available kits (Rel Assay Diagnostics[®]). In TAC assay,
17 antioxidants in the sample reduce dark blue-green colored ABTS radical to colorless reduced ABTS form.
18 The change of absorbance at 660 nm is related to the total antioxidant level of the sample. The assay is
19 calibrated with a stable antioxidant standard solution, Trolox Equivalent, that is a vitamin E analog. Since
20 the separate measurement of different oxidant molecules is not practical and their oxidant effects are
21 additive, the TOS of a sample is measured and is named total peroxide (TP), serum oxidation activity
22 (SOA), reactive oxygen metabolites (ROM) or some other synonyms. In TOC assay, oxidants present in the
23 sample oxidize the ferrous ion-chelator complex to ferric ion. The oxidation reaction is prolonged by
24 enhancing molecules, which are abundant in the reaction medium. The ferric ion makes a colored complex
25 with chromogen in an acidic medium. The color intensity, which can be spectrophotometrically measured,
26 is related to the total amount of oxidant molecules present in the sample (Erel, 2004, 2005).

27

28 2.11. *Statistics and pharmacokinetics analysis*

1 Data are expressed as the mean \pm standard deviation (SD) of three repetitions. One-way analysis of
2 variance (ANOVA) was used to determine the significant differences between the groups followed by a
3 Dunnett post-hoc test for multiple comparisons. Probabilities < 0.05 and < 0.1 were considered as
4 significant. All analyses were performed using SPSS version 15.0 (SPSS Inc[®]).

5

6 **3. Results and discussion**

7 *3.1. Preparation and characterization of SLNs LA-MEM*

8 The solid lipid nanoparticles were made of stearic acid as lipid, Brij 78 (polyoxyethylen(20) stearyl ether)
9 as surfactant, and LA-MEM as drug (Figure 1). In this formulation the stealth character is due to the
10 presence of Brij 78; it was derived from stearic acid molecules covalently grafted with PEG 1000 to form a
11 hydrophilic steric shell around the SLNs. The presence of this hydrophilic PEG chain is expected to protect
12 SLNs from interaction with human serum albumin, to prolong the half-life of SLNs in blood circulation,
13 leading to higher and protracted drug plasma levels *in vivo* (Zhang et al., 2007).

14 LA-MEM loaded SLNs were prepared according to the emulsification-evaporation-solidifying method as
15 described above. The properties, such as intensity average diameter, polydispersity indices, zeta potential,
16 yield, and entrapment efficiency are reported in Table 1. To define the best formulation, LA-MEM loaded
17 SLNs were formulated with different drug:lipid ratio, 1:20, 1:10, 1:5, and 1:2.5, respectively. It was found
18 that the size of LA-MEM loaded SLNs was slightly greater than that of the drug-free ones, due to the
19 incorporation of drug into the particles matrix. The different formulations did not show significant changes
20 in yield, zeta potential, and entrapment efficiency. The entrapment efficiency was evaluated after filtration
21 of the particles with 1.20 μm filters to remove the entrapped drug and, after centrifugation of the particles,
22 to assess the most suitable method to remove the free drug from the system (data not shown). The two
23 methods were not significantly different, and the very similar values of %EE for each type of formulation
24 suggest that the system is not completely saturated, although it was difficult to test higher drug
25 concentrations due to its solubility problems. Thus, the formulation with 1:5 drug:lipid ratio was chosen as
26 the optimized formulation because of the best properties in term of particle size and PDI values (Table 1).

27 We choose 1:0.4 lipid:surfactant ratio after evaluation of different lipid:surfactant ratios. In this case an
28 optimized particle size with a narrow size distribution was obtained, while increasing the amount of Brij 78

1 both the mean particle size and polydispersity index rose (Table A.3). High concentration of surfactant
2 could lead to intrinsic thermodynamic instability of the nanoparticle system causing adsorption of
3 surfactant on the particle surface (Feritas and Muller, 1999; Goppert and Muller, 2005). Likely, at a low
4 concentration surfactant prevents aggregation but, on the contrary, rising the concentration could cause the
5 rearrangement of the surfactant molecules to form loops and tails on the particles surface, leading to the
6 bridging among the SLNs followed by particles aggregation.

7 Chemical and physical instabilities may be observed when the nanoparticles are stored as aqueous
8 suspension (Abdelwahed et al., 2006). In particular, different problems can occur with SLNs, such as lipid
9 crystallization and/or polymorphic transformations, particle growth as a result of Ostwald ripening
10 processes, aggregation/and hydrolysis reactions (Westesen et al., 1997; Muller et al., 2000). For these
11 reasons, nanoparticles are usually subjected to lyophilization process, but this step may generate various
12 stresses resulting in destabilization of the nanoparticles; for example, freezing may lead to crystallization of
13 surface poly(ethylene glycol) (PEG), reducing its steric stabilization properties and leading to particle
14 fusion (Howard et al., 2012). To minimize these stresses and avoid possible particle aggregation and
15 destabilization during the lyophilization process, suitable cryoprotectant were added. In particular, between
16 two tested cryoprotectants we selected sucrose at two different concentrations during the lyophilization
17 process: 5% w/v and 10% w/v aqueous solution (1:10, 1:20 particles:cryoprotectant ratio) of cytoprotectant.
18 As reported in Figure A.1 (Appendices), both dried blank and drug-loaded SLNs, after the reconstitution
19 with deionized water, presented considerable aggregation phenomena, with a greater extent for drug-loaded
20 nanoparticles than the drug-free ones. Higher concentration sucrose (10% w/v) was selected as most
21 suitable compound to maintain the particles size; indeed, it led to a slight increase of the size in the blank
22 nanoparticles (12.9%), although a more considerable raise of particles size in the drug-loaded nanoparticles
23 was observed. From the reported data it is evident the good ability of sucrose as cryoprotectant for this
24 system.

25 The TEM image of SLNs-LA-MEM is shown in Figure 2. It can be observed from these microscopic
26 images that the lipid particles were disk shaped.

27

28 *3.2. Characterization of SLNs-LA-MEM by DSC*

1 The most frequent application of DSC in the study of SLNs dispersions is the investigation of the physical
2 state of the lipid. This is due to the changes of the properties of the matrix lipid materials that can occur
3 during the formation of the colloidal dispersed state. Even when crystalline lipids are used for the
4 preparation of the dispersions, the lipid particles are not necessarily present in the solid state after
5 processing, particularly when the preparation method requires heating. Thus, it is possible to observe
6 supercooling melts, where lipid crystallization may not occur even though the sample is stored at a
7 temperature below the melting point of the lipid. The confirmation of the desired physical state of the
8 matrix lipid is thus of crucial importance for the development of nanoparticles based on solid lipid, as the
9 potential advantages of SLNs are associated to the solid state of the lipid (Bunjés and Unruh, 2007).

10 The calorimetric curve of stearic acid (SA) is characterized by the presence of a small peak at 57.32 °C, and
11 a main peak centred at 69.40 °C due to the melting of the compound; the melting peak of the surfactant is
12 centred at 43.10 °C. Both the blank and the drug-loaded SLNs presented well-defined peaks for lipid and
13 surfactant, but shifted at lower temperatures (66.56 °C and 41.68 °C, 63.01 °C and 36.35 °C for the blank
14 and drug-loaded particles respectively), and slightly broader compared to the bulk materials signals (Figure
15 3). The lowered lipid melting peak of the SLNs suggests that the SA matrix has been successfully solidified
16 during the preparation of the particles, thus, as previously reported by Lee et al. (2007), these results
17 confirm that solid lipid nanoparticles were obtained. The shift of the endothermic signal to lower
18 temperatures could be related to the particle size, according to the Gibbs-Thomson equation: for a particle
19 of given size, it is possible to observe a decrease in melting temperature compared to the bulk material.
20 Certainly, as the nanoparticles dispersions variously polydispersed, the melting transition could also be
21 broadened since the fractions of different particle sizes melt at different temperatures (Lee et al; 2007).

22

23 *3.3. In vitro release and stability studies of SLNs-LA-MEM*

24 The oral route continues to be a challenge, even though it is the most attractive way to administer drugs.
25 After oral administration SLNs may enhance the transport of drugs through the GI tract by different
26 mechanisms including lymphatic transport, drug protection efficiency, and sustained drug release (García-
27 Fuentes et al.; 2005). For this reason, the incorporation of drugs into SLNs opens the perspective of
28 enhanced and/or less variable bioavailability and prolonged plasma levels (Runge, 1998; Demirel et al.,

1 2001). The advantages of SLNs are also linked to their size within the submicron range; therefore, the
2 preservation of the particles size after oral administration is of crucial importance. The first barrier is
3 constituted by the physicochemical environment of the gastrointestinal tract, thus an evaluation of the
4 nanoparticles stability in this medium is necessary. The GI environment may have influence on the physical
5 stability of SLNs formulations due to ionic strength, pH values, and enzymatic activity which destabilize
6 nanoparticles and potentially lead to aggregation.

7 In this study the SLNs stability was investigated, after the incubation of the nanoparticles in the simulated
8 gastrointestinal fluids, by observing potential changes of the particles size distribution and polydispersity
9 index (Alhnan et al., 2011). The first media used were simulated gastric fluids at pH 1.2 with and without
10 pepsin. As shown in Figure 4, the particles seemed stable in both environments, because no significant
11 aggregation was detected by dynamic light scattering. After 2 hours of incubation the size and the PDI were
12 maintained. In literature, some studies have demonstrated the influence of the particles surface on the
13 gastrointestinal stability. Particularly, in the presence of a PEG coating the stability is improved (Tobio et
14 al.; 2000). In this study, the presence of the Brij 78 may, in a similar way, protect particles from
15 degradation and enzymatic attack from pepsin.

16 After demonstrating the stability of SLNs in the stomach, intestinal stability was also studied. Figure 5
17 shows the stability of the particles after incubation in the intestinal simulated fluids at pH 6.8, in the
18 presence or absence of pancreatin. As reported, in both media the particles were subjected to aggregation as
19 well as increase in particle size and PDI. This phenomenon seems to be related to the pH increase of the
20 environment more than to the presence of the enzyme, even if in the latter case the determination of the
21 particle size by dynamic light scattering was particularly compromised by the presence of very large
22 aggregates. Although the question concerning the instability of the particles in the intestine is not fully
23 understood, it is possible that the alkaline pH assists the formation of hydrophobic interactions between the
24 particles and the codrug released in the media, and the Brij 78 coating is not sufficient to protect particles
25 from these interactions. As one of the objectives of the present work was the GI absorption of intact
26 particles and their possibility to reach the brain, the evaluation of the release profile in the simulated
27 gastrointestinal fluids was required. The *in vitro* release studies were conducted in sink conditions in simple
28 SGF and SIF without enzymes as media, at pH 1.2 and 6.8, respectively. Solubility of LA-MEM in

1 simulated gastric and intestinal fluids at room temperature was $<10 \mu\text{g/mL}$; to provide sink conditions, two
2 different concentrations of Cremophor ELP (1% v/v, 0.1% v/v) were tested. Even if 1% (v/v) Cremophor
3 ELP led to a greater solubility increase in the fluids (about $100 \mu\text{g/mL}$), it could not be used because of its
4 ability to form micelles in the media. The dynamic light scattering of the system (SLNs, SGF or SIF with
5 1% v/v of Cremophor) showed a peak of the nanoparticles and a peak related to the presence of micelles at
6 lower values. This did not happen with 0.1% (v/v) of Cremophor in which the solubility was about 60
7 $\mu\text{g/mL}$, allowing it to be selected as final concentration in the media. The presence of common non-ionic
8 surfactant in the release media could also better mimic *in vivo* conditions because it takes into account the
9 large number of surface-active molecules in the body (Wischle and Schwendeman, 2008).

10 The codrug release from SLNs in the SGF was shown in Figures 6-7. The results indicated that in both
11 media, a small burst of release in the first hour was present, followed by a more sustained trend. In
12 particular, in gastric and intestinal simulated fluids after 2 and 4 hours, respectively, approximately only
13 20% of the compound was released. Thus, it is reasonable to assume that the particles may be absorbed
14 before a considerable amount of drug is released in the gut, improving its bioavailability. After the
15 lyophilization of LA-MEM loaded SLNs, a one month stability study was conducted to evaluate the
16 particles size stability after storage at $4 \pm 2 \text{ }^\circ\text{C}$; comparisons were made with SLNs stored as aqueous
17 suspensions. As reported in Figure A.2 (Appendices) LA-MEM loaded SLNs showed a significant increase
18 in particles size in the case of the aqueous suspensions; on the contrary, lyophilized SLNs showed greater
19 consistency in their particles size measurement. Regarding the PDI values, they all remained < 0.3 ,
20 generally considered the acceptable limit of monodispersity. The entrapment efficiency decreased in the
21 suspension after one month storage from 88% to 62%, while overall it remained unchanged after storage
22 for the lyophilized SLNs. The lyophilization process seemed to be effective in terms of nanoparticles long-
23 term stability.

24

25 3.4. Cytotoxicity studies of SLNs-LA-MEM

26 To quantify cell death in response to different SLNs formulations, MTT and LDH assays were performed.
27 The cultured PHWB and N2a NB cells exposed to $10 \mu\text{M}$ of SLNs, SLNs-LA, SLNs-MEM, and SLNs-LA-
28 MEM did not show any significant changes in cell viability during 24 hours, as determined by MTT and

1 LDH assays (data not shown). Likewise, the results of both cytotoxicity assays showed that 100 μ M of
2 SLNs, SLNs-LA, and SLNs-LA-MEM did not cause any significant changes in cell viability, but the
3 application with 100 μ M of SLNs-MEM caused a slight toxicity ($p > 0.05$, $p < 0.1$) compared to the
4 untreated group. Moreover, when the PHWB and N2a NB cells were exposed to 50 μ M H_2O_2 , MTT
5 absorbance values were respectively $56.63 \pm 4.25\%$ and $47.62 \pm 5.32\%$ of that of untreated controls,
6 indicating that H_2O_2 caused cell death. H_2O_2 -induced cell death was also clearly evidenced by about four-
7 and five-folds increases in the LDH activities of positive controls of the cultures (Figures 8-9). The
8 applications with 10 μ M of SLNs, SLNs-MEM, and SLNs-LA-MEM did not lead to any alterations in TAC
9 and TOS levels compared to control value, while 10 μ M concentrations of SLNs-LA and SLNs-LA-MEM
10 supported a slight antioxidant activity ($p > 0.05$) in both cell types without changing TOS levels (data not
11 shown). A concentration of 100 μ M significantly ($p < 0.05$) increased TAC levels in both PHWB and N2a
12 NB cells. Furthermore, TAC and TOS levels were similar in 100 μ M SLNs-LA-MEM treated and untreated
13 cultures, while SLNs-MEM treatments altered the oxidative status. In addition, these alterations were more
14 prominent in N2a NB cells than PHWB cells. Our results also illustrated that H_2O_2 exposure for 24 hours
15 resulted in significant decreases in TAC levels and elevations of TOS levels, compared to those of the
16 control groups. Figures 10-11 show the *in vitro* effects of different formulations of SLNs at 100 μ M on
17 TAC and TOS levels, respectively.

18 It has been reported that the benefit-risk balance deriving from the use of NPs for the treatment of CNS
19 diseases should be carefully evaluated for each type of new engineered NP intended for brain-specific drug
20 delivery (Re et al., 2012). In this study the results of our cytotoxicity assays (MTT and LDH) showed that
21 even 100 μ M of SLNs alone did not cause any significant changes in cell viability and oxidative status. In
22 parallel to our findings, recent reports indicated that SLNs could be used as safe and efficient delivery
23 systems for brain since they had no toxicity (Blasi et al.; 2013). On the contrary, in the present investigation
24 the application of 100 μ M of SLNs-MEM induced a slight cytotoxicity leading to a reduction of cell
25 viability likely related to oxidative alterations in N2a NB and PHWB cells. Similar to these findings, MEM
26 applications at 100 μ M led to notable decreases of cell viability in both N2a NB and MDCK cell lines, as
27 determined by MTT assay during 24 h (Coleman et al.; 2013). Again, the determined TAC and TOS levels
28 were similar in 100 μ M SLNs-LA-MEM treated and untreated cultures while SLNs-MEM treatments

1 altered the oxidative status. In fact, antioxidant-loaded SLN systems - containing resveratrol, quercetin,
2 idebenone, and ferulic acid, respectively - were found to be effective in decreasing ROS accumulation, thus
3 exerting antioxidant activity in both *in vitro* and *in vivo* experimental models (Dhawan et al.; 2011). For the
4 first time, cytotoxicity and oxidative damage analyses revealed that SLNs-LA-MEM could be a suitable
5 DDS for an *in vivo* examination for the brain delivery of LA-MEM codrug which could be able to mitigate
6 the oxidative damage and increase the antioxidant capacity in many neurodegenerative disorders,
7 particularly in AD.

8

9 **4. Conclusion**

10 The new nanoparticulate delivery system loaded with LA-MEM was successfully prepared and
11 characterized in terms of physicochemical properties, entrapment efficiency, and release profile in
12 simulated gastrointestinal fluids. The optimized formulation (1000 rpm, 1:5 drug:lipid ratio and 0.03% of
13 surfactant) had an average intensity diameter of 170 nm, a PDI of 0.072, a zeta potential of -33.8 mV, and
14 an entrapment efficiency of 88%, in distilled water. Also the data obtained from our preliminary *in vitro*
15 biological tests clearly revealed that SLNs-LA-MEM could be screened as a suitable DDS for the brain,
16 from a toxicological point of view. Our preliminary results suggest the suitability of these nanoparticulate
17 devices as carrier systems for LA-MEM codrug for further *in vivo* studies.

18

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21 **Disclosure**

22 The author reports no conflicts of interest in this work.

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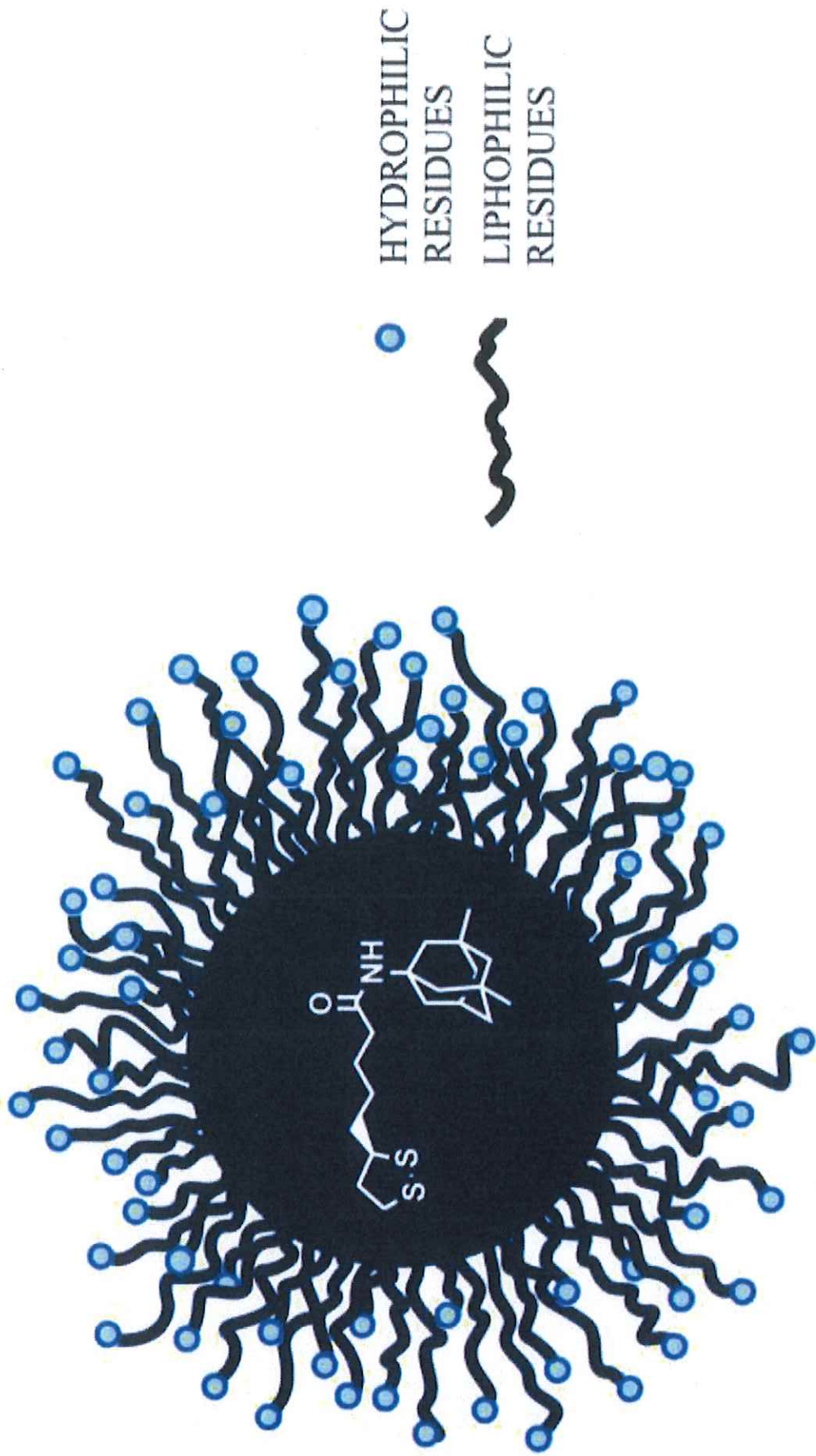
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36



Supplementary Material

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Tables and Figures

Solid Lipid Nanoparticles Loaded With Lipoyl-Memantine Codrug:

Preparation and Characterization

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Table 1 Drug Loading Capacity (DLC) and Entrapment Efficiency (EE) of several ratios of LA-MEM loaded SLN.

LA-MEM loaded SLN Codrug:Lipid Ratio	Size (nm)	PdI	ZP (mV)	Recovery (%)	EE (%)	DLC (%)
Blank	160.0 ± 1.2	0.099 ± 0.01	-27.4 ± 0.8	95.50 ± 1.3	-	-
1:2.5	195.6 ± 1.6	0.090 ± 0.01	-33.90 ± 1.05	95 ± 2.8	92 ± 2.8	-
1:5	169.9 ± 4.3	0.072 ± 0.01	-33.8 ± 0.9	96.0 ± 1.7	88 ± 1.8	12.5 ± 1.3
1:10	183.05 ± 1.9	0.120 ± 0.01	-32.9 ± 2.4	97.5 ± 0.7	90 ± 1	-
1:20	173.5 ± 1.4	0.084 ± 0.01	-35 ± 2.1	97.0 ± 0	90 ± 0.7	-

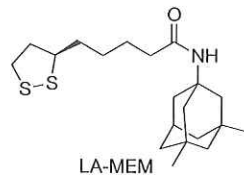


Fig. 1. Chemical structure of LA-MEM codrug.

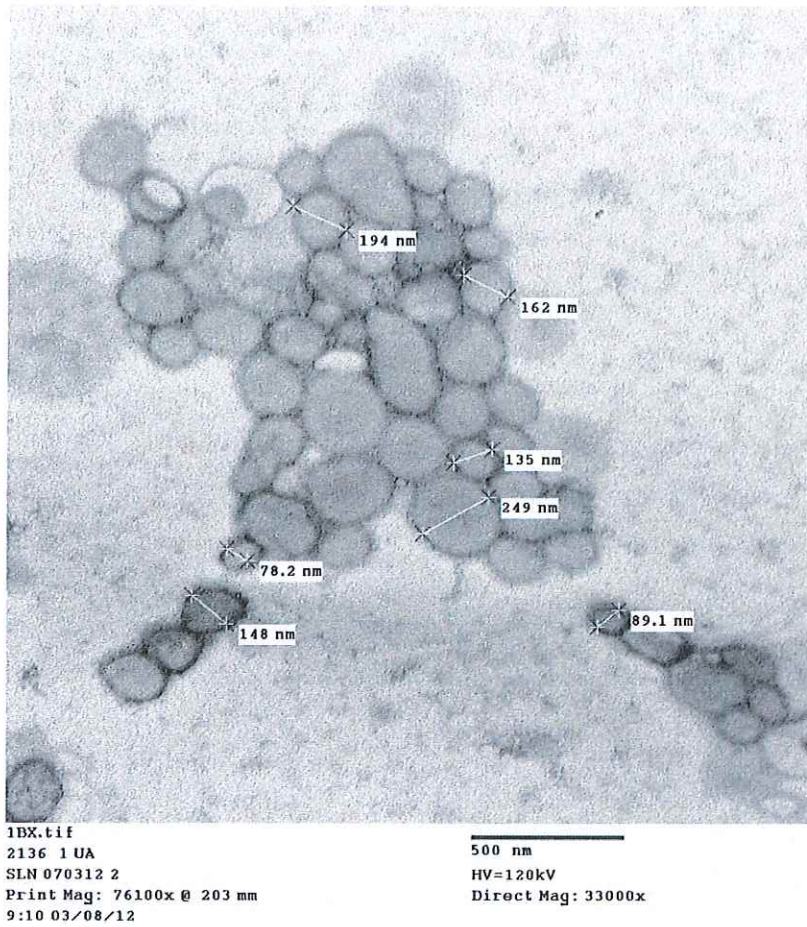


Fig. 2. TEM image of LA-MEM loaded SLN.

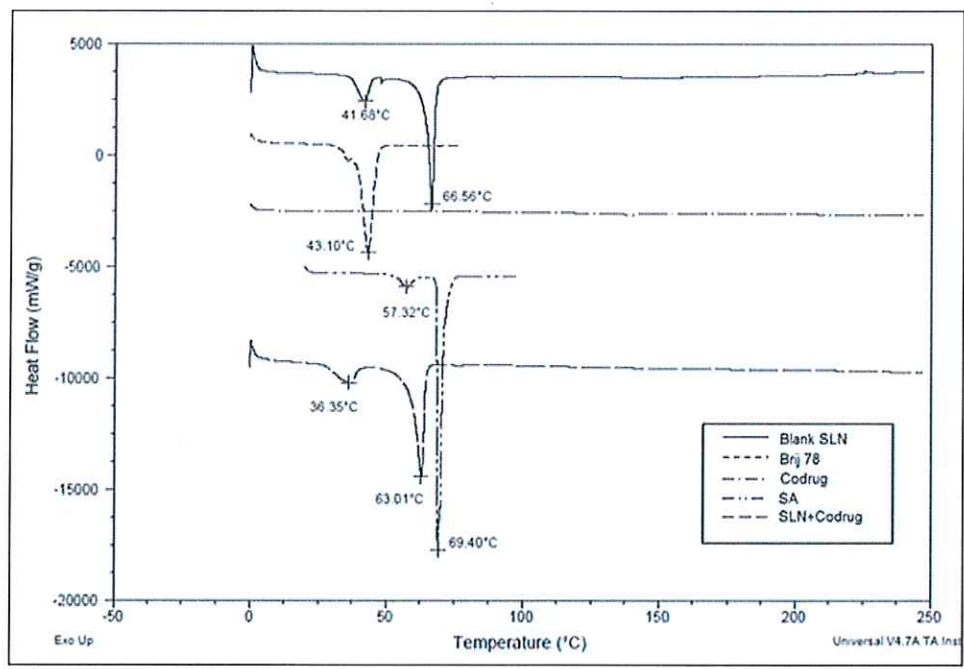


Fig. 3. Differential scanning calorimetry scans of LA-MEM loaded SLN.

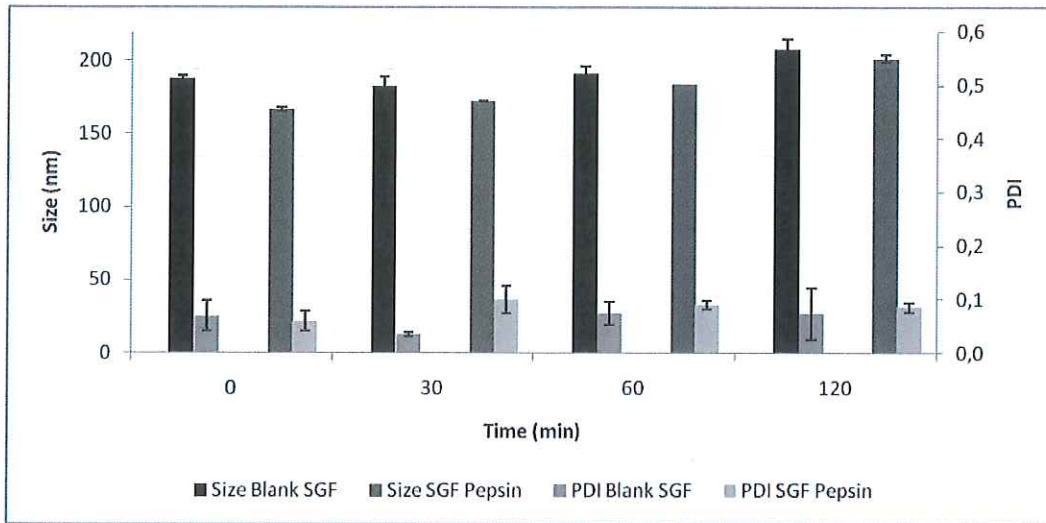


Fig. 4. Size and PDI of SLNs-LA-MEM after incubation in SGF (pH 1.2) at 37 °C, 100 rpm.

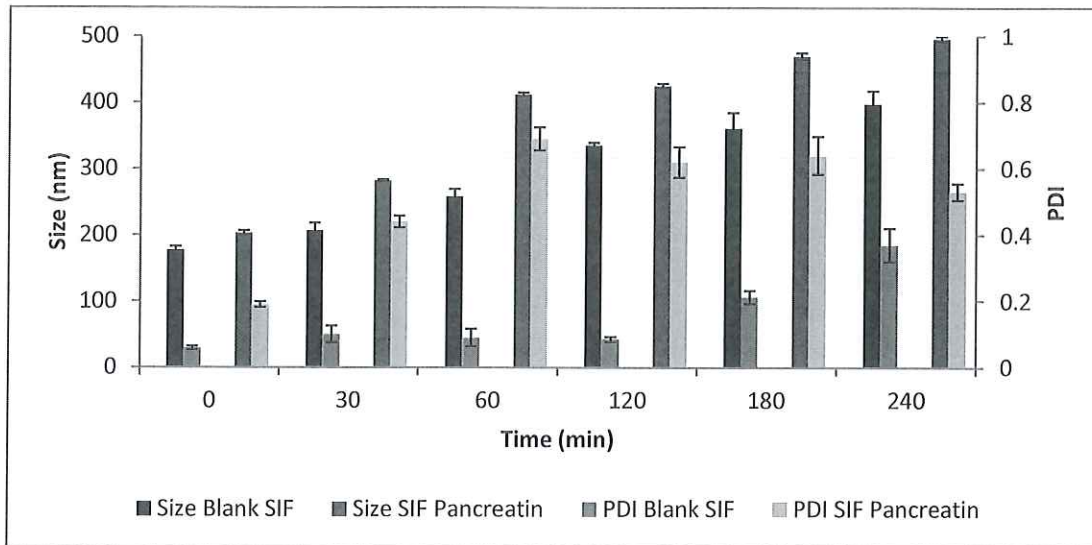


Fig. 5. Size and PDI of SLNs-LA-MEM after incubation in SIF (pH 6.8) at 37 °C,100 rpm.

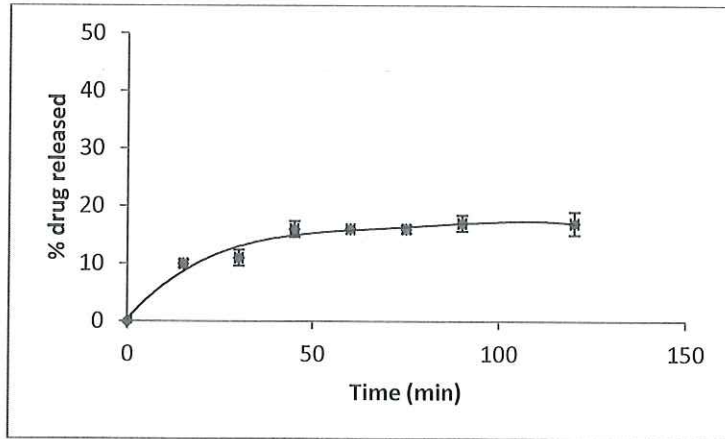


Fig. 6. *In vitro* release of LA-MEM from SLNs after incubation in SGF (pH 1.2) and 0.1% (v/v) Cremophor ELP, at 37 °C, 100 rpm.

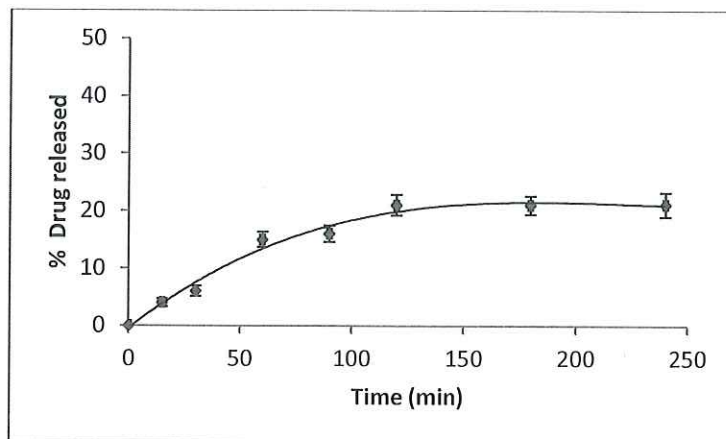


Fig. 7. *In vitro* release of LA-MEM from SLNs after incubation in SIF (pH 6.8) and 0.1% (v/v).

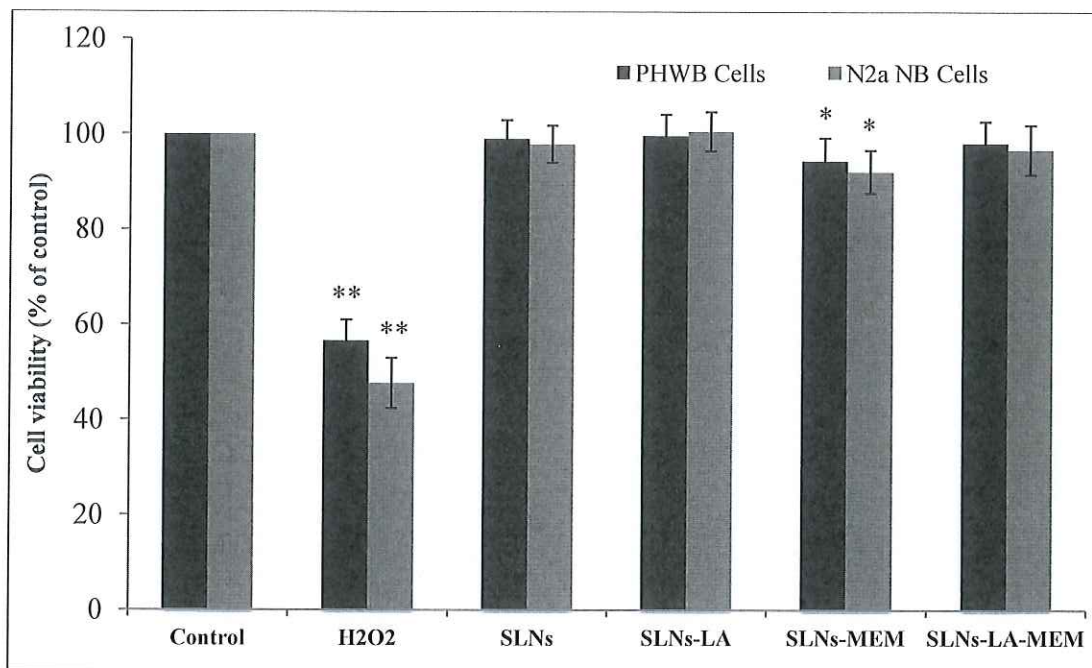


Fig. 8. Viability of PHWB and N2a NB cells after 24 h exposure to (100 μ M) different SLNs formulations. The results were presented as percentage of the control group (n = 3). Data are expressed as mean \pm SD. ** p < 0.05, * p < 0.1.

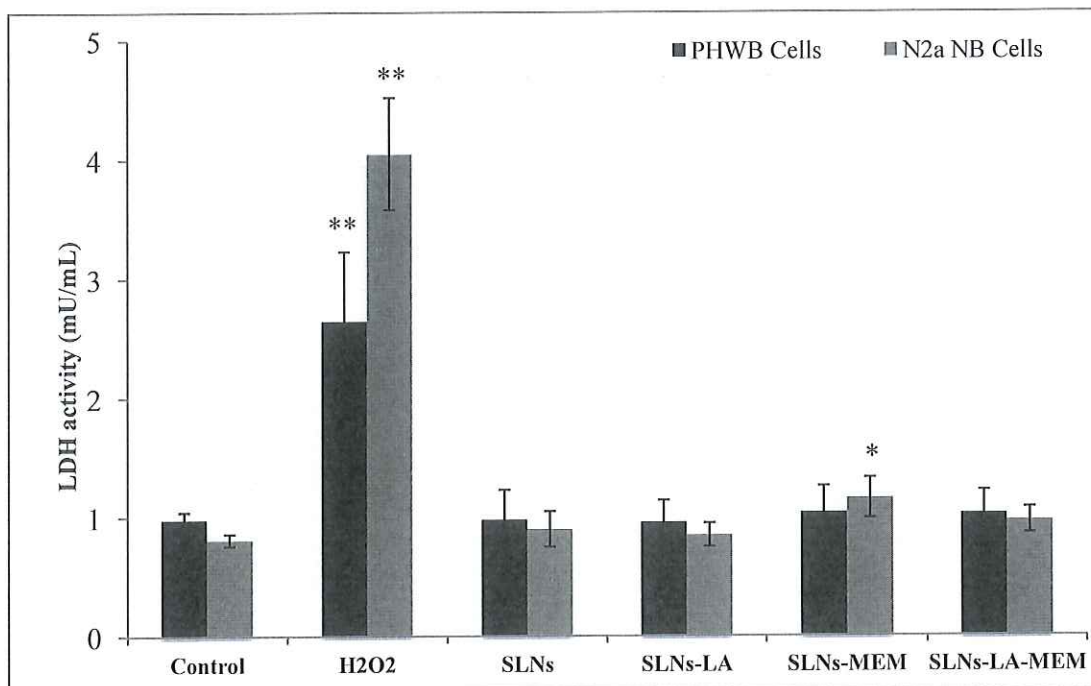


Fig. 9. Extracellular levels of LDH in cultured PHWB and N2a NB cells maintained in the presence of different SLNs formulations (100 μ M).

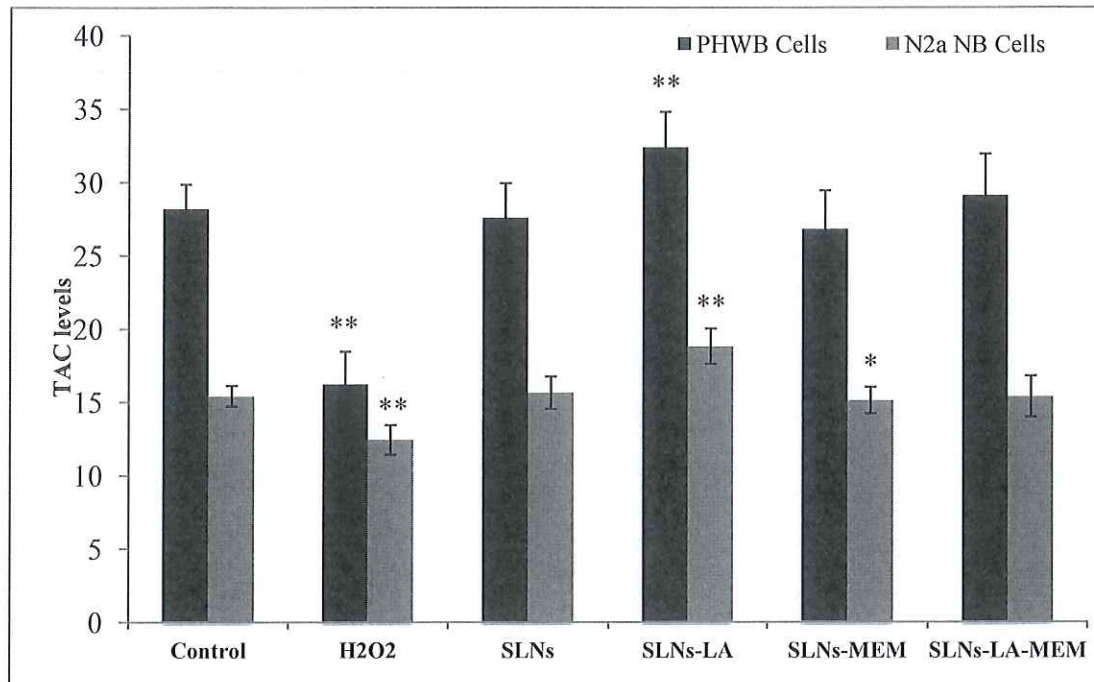


Fig. 10. *In vitro* levels of TAC (as mmol Trolox Equiv./L) in cultured PHWB and N2a NB cells maintained in the presence of different SLNs formulations (100 μ M).

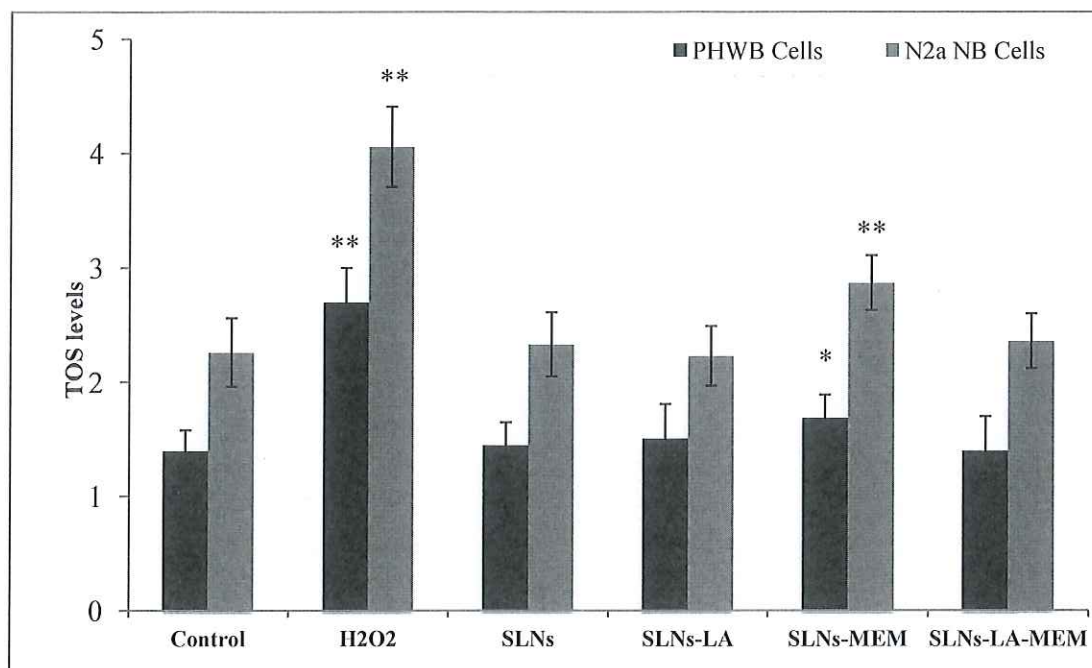


Fig. 11. *In vitro* levels of TOS (as mmol H₂O₂ Equiv./L) in cultured PHWB and N2a NB cells maintained in the presence of different SLNs formulations (100 μM).

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