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Corrigendum

Corrigendum to "Low-cost temperature transition mixtures (TTM) based on ethylene glycol/potassium hydroxide as reversible CO_2 sorbents" [J. Mol. Liquids 340 (2021) 117180]

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In this corrigendum, we report a correction on the calculated values of activation energies both of viscosity (E_η) and of ionic conductivity (E_σ) reported in the paper: Journal of Molecular Liquids 340 (2021) 117180.

In the paper "Low-cost temperature transition mixtures (TTM) based on ethylene glycol/potassium hydroxide as reversible CO_2 sorbents" (https://doi.org/10.1016/j.molliq.2021.117180) we measured viscosity, density and conductivity for two systems based on ethylene glycol (EG), potassium hydroxide and boric acid (BA).

The experimental values and the Walden plots have been checked and resulted reproducible, but unfortunately, we found a mis-calculation on the values of E_η and E_σ. In this article we report the correct values (Table 1).

The values reported in the paper Journal of Molecular Liquids 340 (2021) 117180 are lower, these are correct and coherent with literature data on similar structured liquids [1].

The considerations made in the discussion of the paper are still valid as the values are lower to the ones reported for similar liquids and similar to other low-viscous systems [2], therefore mainly determining the physical absorption phenomena of CO_2 in these liquids, as stated in the main paper.

The authors would like to apologise for any inconvenience caused.

References

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Table 1
Viscosity Activation Energies (E_η, kJ/mol) and Ionic Conductivity Activation Energies (E_σ, kJ/mol) for System A (EG/KOH 3:1) and System B (EG/KOH/BA 3:1:1) liquids.

	Viscosity Activation Energy E_η (kJ/mol)	Ionic Conductivity Activation Energy E_σ (kJ/mol)
System A (EG/KOH 3:1)	43.20	26.30
System B (EG/KOH/BA 3:1:1)	36.86	20.39