

# Mononuclear Rearrangement of Heterocycles in Zwitterionic Micelles of Amine Oxide Surfactants

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## Supporting Material

Figure S1. Dependence of the observed rate constant ( $k_{\text{obs}}$ , s<sup>-1</sup>) for the MRH **1a** → **2a** on the concentration of C<sub>14</sub> DMAO.

Figure S2. Dependence of the observed rate constant ( $k_{\text{obs}}$ , s<sup>-1</sup>) for the MRH **1b** → **2b** on the concentration of C<sub>14</sub> DMAO.

Figure S3. Dependence of the observed rate constant ( $k_{\text{obs}}$ , s<sup>-1</sup>) for the MRH **1c** → **2c** on the concentration of C<sub>14</sub> DMAO.

Figure S4. Dependence of the observed rate constant ( $k_{\text{obs}}$ , s<sup>-1</sup>) for the MRH **1d** → **2d** on the concentration of C<sub>14</sub> DMAO.

Figure S5. Dependence of the observed rate constant ( $k_{\text{obs}}$ , s<sup>-1</sup>) for the MRH **1e** → **2e** on the concentration of C<sub>14</sub> DMAO.

Table S1. C.m.c. values for C<sub>14</sub> DMAO in the presence of the investigated substrates.

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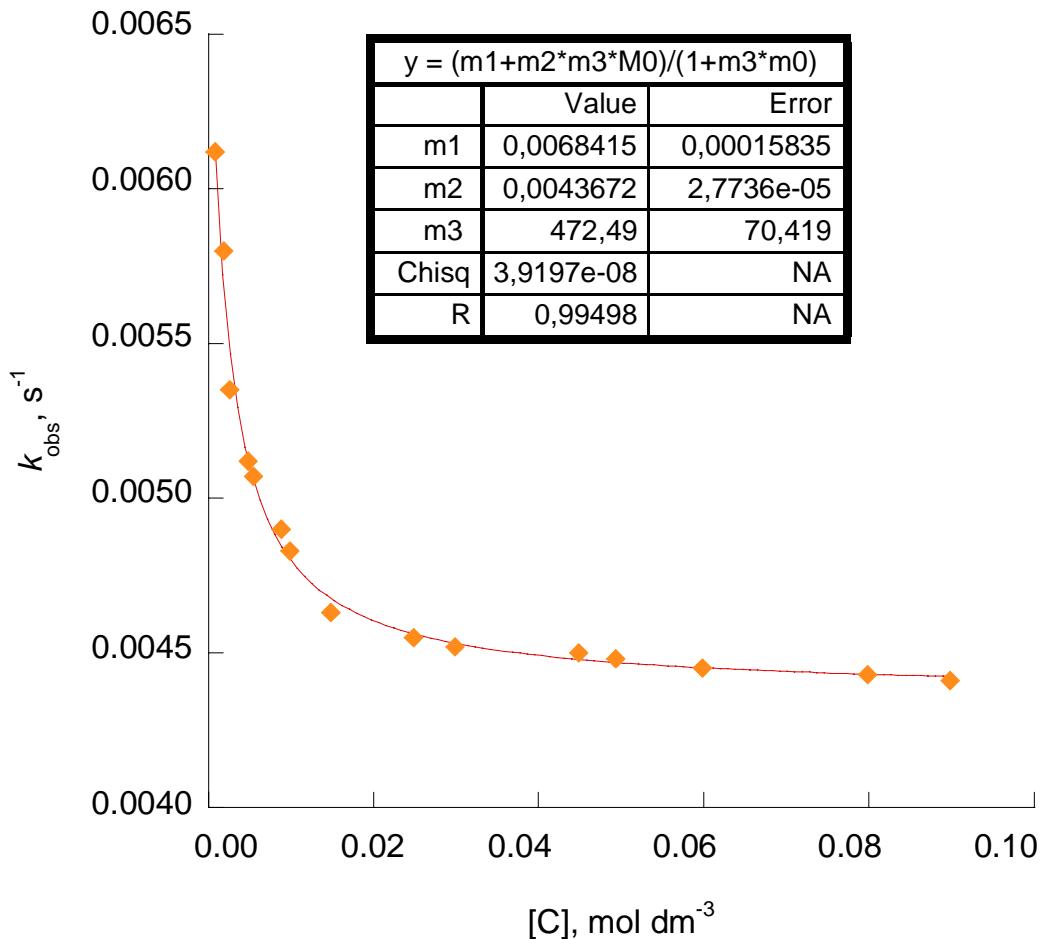


Figure S1. Dependence of the observed rate constant ( $k_{\text{obs}}$ ,  $\text{s}^{-1}$ ) for the MRH **1a**  $\rightarrow$  **2a** on the concentration of C<sub>14</sub> DMAO (concentration corrected for the corresponding c.m.c. of the surfactant,  $1.5 \times 10^{-4}$  mol  $\text{dm}^{-3}$ ) in the presence of borate buffer ( $\text{pH}$  9.6), ionic strength  $2.0 \times 10^{-2}$  mol  $\text{dm}^{-3}$ , at  $313.1 \pm 0.1$  K.

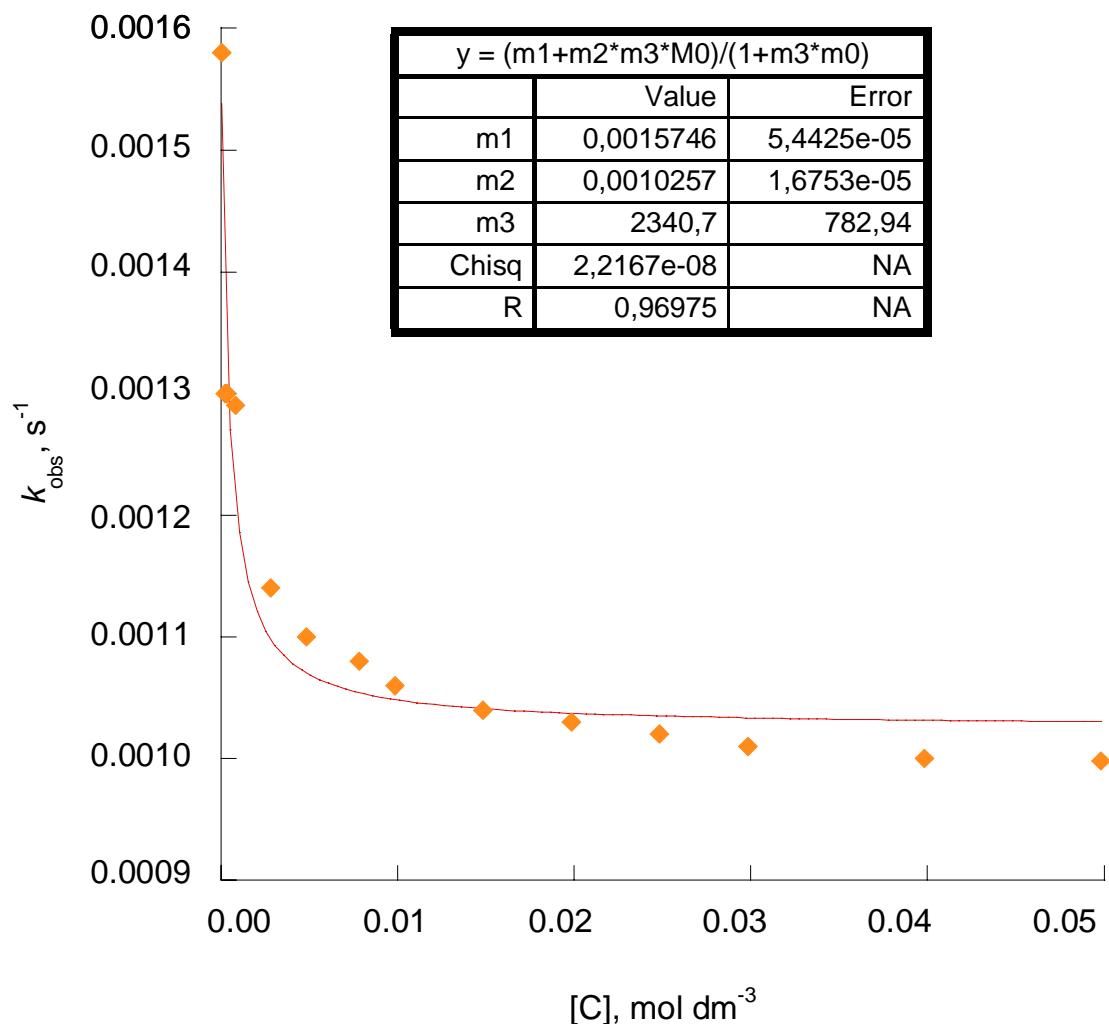


Figure S2. Dependence of the observed rate constant ( $k_{\text{obs}}$ ,  $\text{s}^{-1}$ ) for the MRH **1b** → **2b** on the concentration of  $\text{C}_{14}\text{DMAO}$  (concentration corrected for the corresponding c.m.c. of the surfactant,  $1.7 \times 10^{-4}$  mol  $\text{dm}^{-3}$ ) in the presence of borate buffer ( $\text{pH } 9.6$ ), ionic strength  $2.0 \times 10^{-2}$  mol  $\text{dm}^{-3}$ , at  $313.1 \pm 0.1$  K.

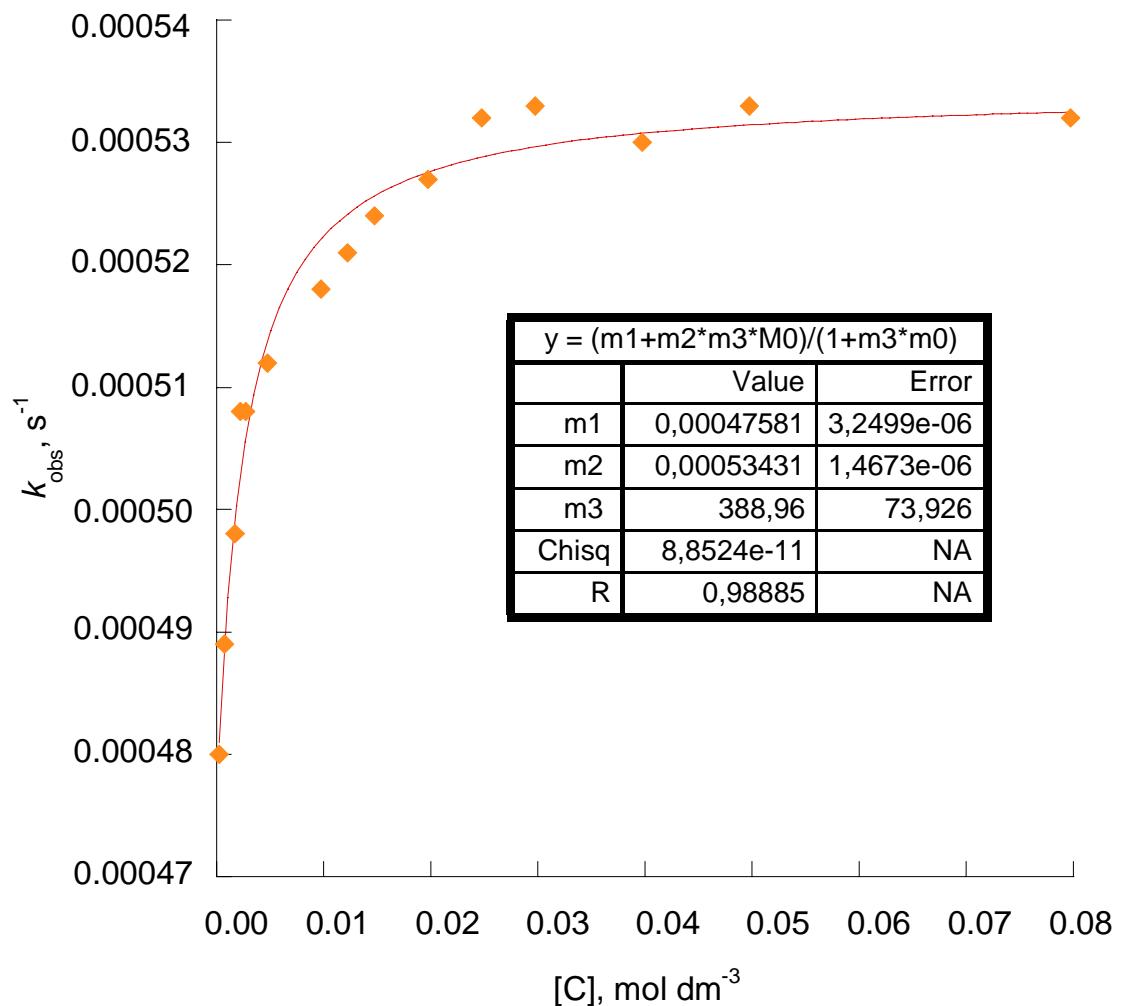


Figure S3. Dependence of the observed rate constant ( $k_{\text{obs}}$ ,  $\text{s}^{-1}$ ) for the MRH **1c** → **2c** on the concentration of C<sub>14</sub> DMAO (concentration corrected for the corresponding c.m.c. of the surfactant,  $2.5 \times 10^{-4}$  mol  $\text{dm}^{-3}$ ) in the presence of borate buffer (pH 9.6), ionic strength  $2.0 \times 10^{-2}$  mol  $\text{dm}^{-3}$ , at  $313.1 \pm 0.1$  K.

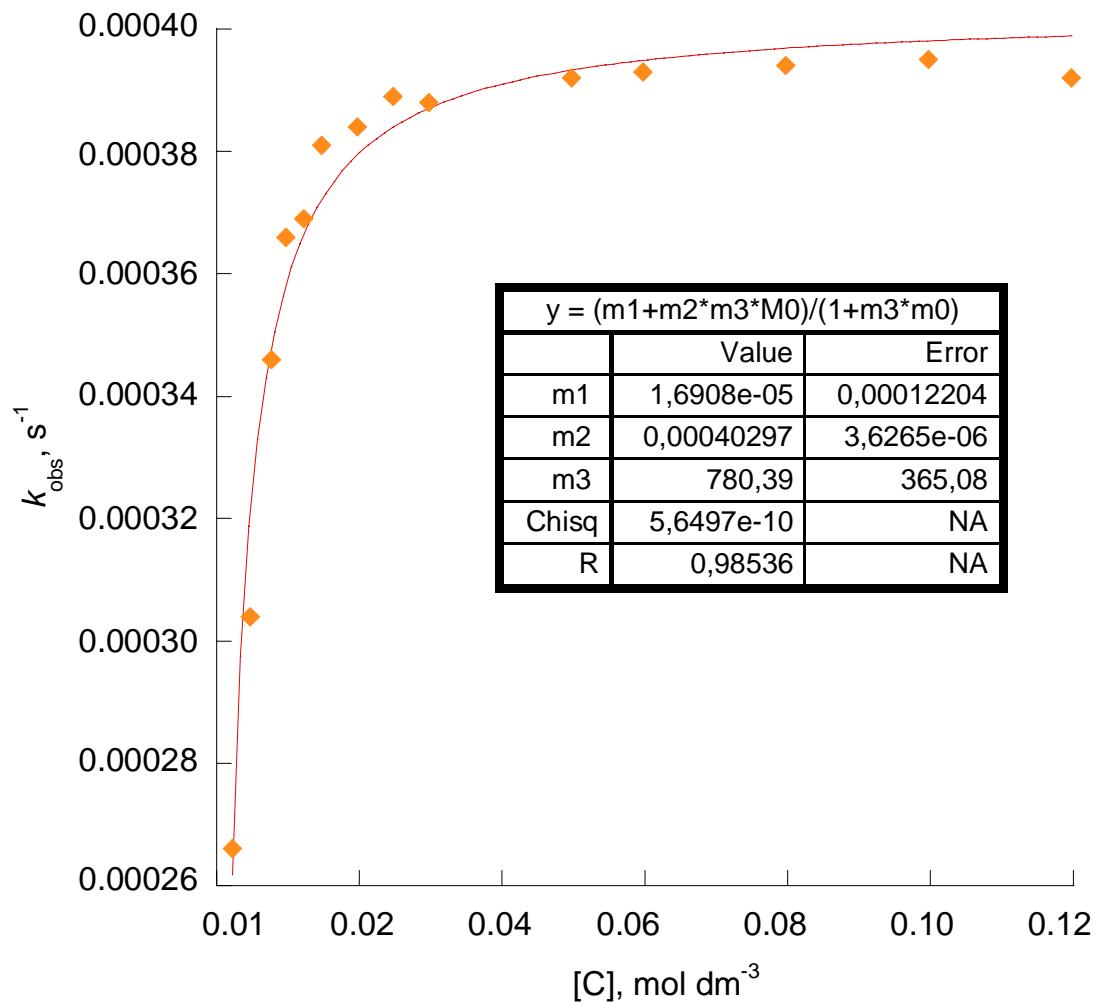


Figure S4. Dependence of the observed rate constant ( $k_{\text{obs}}$ ,  $\text{s}^{-1}$ ) for the MRH **1d** → **2d** on the concentration of  $\text{C}_{14}$  DMAO (concentration corrected for the corresponding c.m.c. of the surfactant,  $2.8 \times 10^{-4}$   $\text{mol dm}^{-3}$ ) in the presence of borate buffer ( $\text{pH } 9.6$ ), ionic strength  $2.0 \times 10^{-2}$   $\text{mol dm}^{-3}$ , at  $313.1 \pm 0.1$  K.

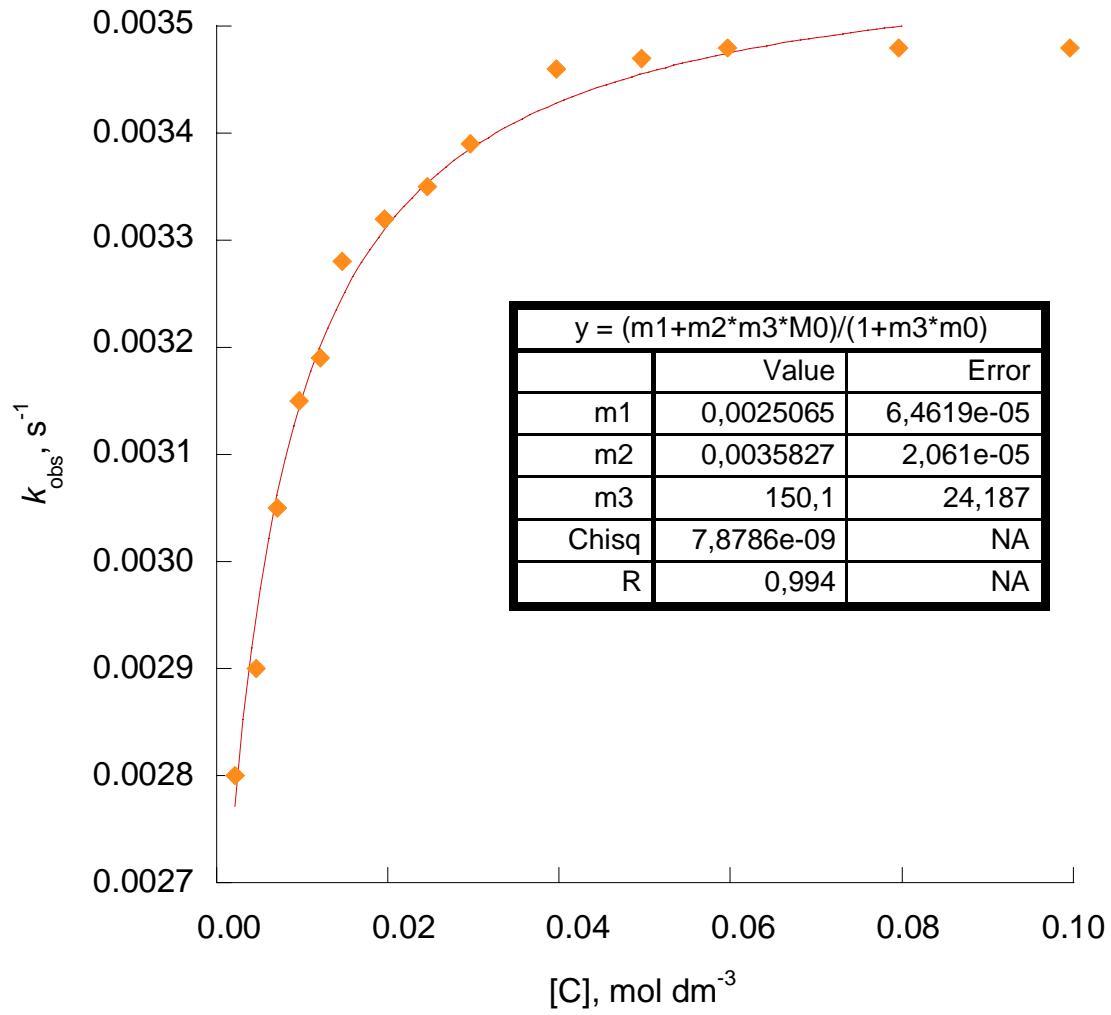


Figure S5. Dependence of the observed rate constant ( $k_{\text{obs}}$ ,  $\text{s}^{-1}$ ) for the MRH **1e**  $\rightarrow$  **2e** on the concentration of C<sub>14</sub> DMAO (concentration corrected for the corresponding c.m.c. of the surfactant,  $3.3 \times 10^{-4}$   $\text{mol dm}^{-3}$ ) in the presence of borate buffer (pH 9.6), ionic strength  $2.0 \times 10^{-2}$   $\text{mol dm}^{-3}$ , at  $313.1 \pm 0.1$  K.

Table S1. C.m.c. values for C<sub>14</sub>DMAO in the presence of the investigated substrates.

C.m.c. (/10 <sup>4</sup> , mol dm <sup>-3</sup> )	
<b>1a</b>	1.5
<b>1b</b>	1.7
<b>1c</b>	2.5
<b>1d</b>	2.8
<b>1e</b>	3.3